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<b>UTILITY PATENT APPLICATION TRANSMITTAL</b> <small>(Only for new nonprovisional applications under 37 C.F.R. § 1.53(b))</small>	Attorney Docket No.	3357
	First Inventor or Application Identifier	McGALL
	Title	SYNTHESIS OF OLIGONUCLEOTIDE, ETC.
	Express Mail Label No.	EL675507530US

APPLICATION ELEMENTS <small>See MPEP chapter 600 concerning utility patent application contents.</small>	ADDRESS TO: Assistant Commissioner for Patents Box Patent Application Washington, DC 20231	
1. <input checked="" type="checkbox"/> * Fee Transmittal Form (e.g., PTO/SB/17) (Submit an original and a duplicate for fee processing)	5. <input type="checkbox"/> Microfiche Computer Program (Appendix)	
2. <input checked="" type="checkbox"/> Specification [Total Pages 21] (preferred arrangement set forth below) <ul style="list-style-type: none"><li>- Descriptive title of the Invention</li><li>- Cross References to Related Applications</li><li>- Statement Regarding Fed sponsored R &amp; D</li><li>- Reference to Microfiche Appendix</li><li>- Background of the Invention</li><li>- Brief Summary of the Invention</li><li>- Brief Description of the Drawings (if filed)</li><li>- Detailed Description</li><li>- Claim(s)</li><li>- Abstract of the Disclosure</li></ul>	6. Nucleotide and/or Amino Acid Sequence Submission (if applicable, all necessary) <ul style="list-style-type: none"><li>a. <input type="checkbox"/> Computer Readable Copy</li><li>b. <input type="checkbox"/> Paper Copy (identical to computer copy)</li><li>c. <input type="checkbox"/> Statement verifying identity of above copies</li></ul>	
3. <input checked="" type="checkbox"/> Drawing(s) (35 U.S.C. 113) [Total Sheets 10]	<b>ACCOMPANYING APPLICATION PARTS</b> 7. <input type="checkbox"/> Assignment Papers (cover sheet & document(s)) 8. <input type="checkbox"/> 37 C.F.R. § 3.73(b) Statement of Power of Attorney (when there is an assignee) <input type="checkbox"/> Attorney 9. <input type="checkbox"/> English Translation Document (if applicable) 10. <input type="checkbox"/> Information Disclosure Statement (IDS)/PTO-1449 <input type="checkbox"/> Copies of IDS Citations 11. <input type="checkbox"/> Preliminary Amendment 12. <input type="checkbox"/> Return Receipt Postcard (MPEP 503) (Should be specifically itemized) 13. <input type="checkbox"/> * Small Entity Statement(s) filed in prior application, Status still proper and desired (PTO/SB/09-12) 14. <input type="checkbox"/> Certified Copy of Priority Document(s) (if foreign priority is claimed) 15. <input type="checkbox"/> Other: .....	
4. Oath or Declaration [Total Pages ] <ul style="list-style-type: none"><li>a. <input type="checkbox"/> Newly executed (original or copy)</li><li>b. <input type="checkbox"/> Copy from a prior application (37 C.F.R. § 1.63(d)) (for continuation/divisional with Box 16 completed)<ul style="list-style-type: none"><li>i. <input type="checkbox"/> <b>DELETION OF INVENTOR(S)</b> Signed statement attached deleting inventor(s) named in the prior application, see 37 C.F.R. §§ 1.63(d)(2) and 1.33(b).</li></ul></li></ul>		
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17. <b>CORRESPONDENCE ADDRESS</b>	
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Name (Print/Type)	Philip L. McGarrigle	Registration No. (Attorney/Agent)	31,395
Signature	<i>Philip McGarrigle</i>	Date	9/11/00

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See 37 C.F.R. §§ 1.27 and 1.28.*

**TOTAL AMOUNT OF PAYMENT** (\$) 852.00

## Complete if Known

Application Number	TBD
Filing Date	9/11/00
First Named Inventor	McGALL
Examiner Name	TBD
Group / Art Unit	TBD
Attorney Docket No.	3357

## METHOD OF PAYMENT (check one)

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Deposit Account Name: Affymetrix, Inc.

- ☒ Charge Any Additional Fee Required Under 37 CFR §§ 1.16 and 1.17

2. ☐ Payment Enclosed:

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## FEE CALCULATION

### 1. BASIC FILING FEE

Large Entity Code	Small Entity Code	Fee (\$)	Fee Description	Fee Paid	
101	690	201	345	Utility filing fee	690
106	310	206	155	Design filing fee	
107	480	207	240	Plant filing fee	
108	690	208	345	Reissue filing fee	
114	150	214	75	Provisional filing fee	

**SUBTOTAL (1)** (\$) 690

### 2. EXTRA CLAIM FEES

Total Claims	Extra Claims	Fee from below	Fee Paid
29	-20** = 9	18	162
3	-3** = 0	0	0
Multiple Dependent			0

\*\*or number previously paid, if greater; For Reissues, see below

Large Entity Code	Small Entity Code	Fee (\$)	Fee Description
103	18	203	9
102	78	202	39
104	260	204	130
109	78	209	39
110	18	210	9

**SUBTOTAL (2)** (\$) 162

## FEE CALCULATION (continued)

### 3. ADDITIONAL FEES

Large Entity Code	Small Entity Code	Fee (\$)	Fee Description	Fee Paid	
105	130	205	65	Surcharge - late filing fee or oath	
127	50	227	25	Surcharge - late provisional filing fee or cover sheet	
139	130	139	130	Non-English specification	
147	2,520	147	2,520	For filing a request for reexamination	
112	920*	112	920*	Requesting publication of SIR prior to Examiner action	
113	1,840*	113	1,840*	Requesting publication of SIR after Examiner action	
115	110	215	55	Extension for reply within first month	
116	380	216	190	Extension for reply within second month	
117	870	217	435	Extension for reply within third month	
118	1,360	218	680	Extension for reply within fourth month	
128	1,850	228	925	Extension for reply within fifth month	
119	300	219	150	Notice of Appeal	
120	300	220	150	Filing a brief in support of an appeal	
121	260	221	130	Request for oral hearing	
138	1,510	138	1,510	Petition to institute a public use proceeding	
140	110	240	55	Petition to revive - unavoidable	
141	1,210	241	605	Petition to revive - unintentional	
142	1,210	242	605	Utility issue fee (or reissue)	
143	430	243	215	Design issue fee	
144	580	244	290	Plant issue fee	
122	130	122	130	Petitions to the Commissioner	
123	50	123	50	Petitions related to provisional applications	
126	240	126	240	Submission of Information Disclosure Stmt	
581	40	581	40	Recording each patent assignment per property (times number of properties)	
146	690	246	345	Filing a submission after final rejection (37 CFR § 1.129(a))	
149	690	249	345	For each additional invention to be examined (37 CFR § 1.129(b))	

Other fee (specify) \_\_\_\_\_

Other fee (specify) \_\_\_\_\_

\* Reduced by Basic Filing Fee Paid

**SUBTOTAL (3)** (\$) 0

## SUBMITTED BY

Name (Print/Type)	Philip L. McGarrigle	Registration No. (Attorney/Agent)	31,395	Telephone	(408) 731-5021
Signature		Date	9/11/00		

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**PATENT APPLICATION**

**SYNTHESIS OF OLIGONUCLEOTIDE ARRAYS USING PHOTOCLEAVABLE  
PROTECTING GROUPS**

**Inventors:**

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Canada, residing at 1121 Sladky Ave.  
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Santa Clara CA 95051**

**Entity:       Large**

1                   **SYNTHESIS OF OLIGONUCLEOTIDE ARRAYS USING**  
2                   **PHOTOCLEAVABLE PROTECTING GROUPS**

3  
4                   **BACKGROUND OF THE INVENTION**

5                   The present invention relates to the area of chemical synthesis. More  
6 particularly, this invention relates to photolabile compounds, reagents for preparing the same  
7 and methods for their use as photocleavable linkers and protecting groups, particularly in the  
8 synthesis of high density molecular arrays on solid supports.

9                   The use of a photolabile molecule as a linker to couple molecules to solid  
10 supports and to facilitate the subsequent cleavage reaction has received considerable  
11 attention during the last two decades. Photolysis offers a mild method of cleavage which  
12 complements traditional acidic or basic cleavage techniques. *See, e.g.*, Lloyd-Williams et al.  
13 (1993) *Tetrahedron* **49**:11065-11133. The rapidly growing field of combinatorial organic  
14 synthesis (*see, e.g.*, Gallop et al. (1994) *J. Med. Chem.* **37**:1233-1251; and Gordon et al.  
15 (1994) *J. Med. Chem.* **37**:1385-1401) involving libraries of peptides and small molecules has  
16 markedly renewed interest in the use of photolabile linkers for the release of both ligands  
17 and tagging molecules.

18                  A variety of *ortho*-benzyl compounds as photolabile protecting groups have  
19 been used in the course of optimizing the photolithographic synthesis of both peptides (*see*  
20 Fodor et al. (1994) *Science* **251**:767-773) and oligonucleotides (*see* Pease et al. *Proc. Natl.*  
21 *Acad. Sci. USA* **91**:5022-5026). *See* PCT patent publication Nos. WO 90/15070, WO  
22 92/10092, and WO 94/10128; see also U.S. patent application Serial No. 07/971,181, filed  
23 2 Nov. 1992, and Serial No. 08/310,510, filed September 22, 1994; Holmes et al. (1994) in  
24 *Peptides: Chemistry, Structure and Biology (Proceedings of the 13th American Peptide*  
25 *Symposium)*; Hodges et al. Eds.; ESCOM: Leiden; pp. 110-12, each of these references is  
26 incorporated herein by reference for all purposes. Examples of these compounds included  
27 the 6-nitroveratryl derived protecting groups, which incorporate two additional alkoxy  
28 groups into the benzene ring. Introduction of an  $\alpha$ -methyl onto the benzylic carbon  
29 facilitated the photolytic cleavage with > 350 nm UV light and resulted in the formation of a  
30 nitroso-ketone.

31                  Photocleavable protecting groups and linkers should be stable to a variety of  
32 reagents (*e.g.*, piperidine, TFA, and the like); be rapidly cleaved under mild conditions; and

not generate highly reactive byproducts. The present invention provides such protecting groups and methods for their use in synthesizing high density molecular arrays.

#### SUMMARY OF THE INVENTION

According to a first aspect of the invention, novel compounds are provided which are useful for providing protecting groups in chemical synthesis, preferably in the solid phase synthesis of oligonucleotides and polypeptides. These compounds are generally photolabile and comprise protecting groups which can be removed by photolysis to unmask a reactive group. The compounds have the general formulas as shown in Figure 1 and 9.

Another aspect of this invention provides a method of attaching a molecule with a reactive site to a support comprising the steps of:

- (a) providing a support with a reactive site;
- (b) binding a molecule to the reactive site, the molecule comprising a masked reactive site attached to a photolabile protecting group of the formula as shown in Figure 1, and
- (c) removing the photolabile protecting group to provide a derivatized support comprising the molecule with an unmasked reactive site immobilized thereon.

A related aspect of this invention provides a method of forming, from component molecules, a plurality of compounds on a support, each compound occupying a separate region of the support, said method comprising the steps of:

- (a) activating a region of the support;
- (b) binding a molecule to the region, said molecule comprising a masked reactive site linked to a photolabile protecting group of the formula as shown in Figure 1;
- (c) repeating steps (a) and (b) on other regions of the support whereby each of said other regions has bound thereto another molecule comprising a masked reactive site linked to the photolabile protecting group, wherein said another molecule may be the same or different from that used in step (b);
- (d) removing the photolabile protecting group from one of the molecules bound to one of the regions of the support to provide a region bearing a molecule with an unmasked reactive site;
- (e) binding an additional molecule to the molecule with an unmasked reactive site;

(f) repeating steps (d) and (e) on regions of the support until a desired plurality of compounds is formed from the component molecules, each compound occupying separate regions of the support.

This method finds particular utility in synthesizing high density arrays of nucleic acids on solid supports in either the 3'→5' or 5'→3' directions.

### **BRIEF DESCRIPTION OF THE DRAWINGS**

Figure 1 shows a general outline of the alternative synthesis chemistries and outlines what the general structures for "Y" could be.

Figure 2 shows specific compounds that are preferred within the general structures shown in Fig. 1 and shows the stepwise yield when they were used to couple nucleotides together and the specific photolysis conditions used..

Figure 3 shows the synthesis of 5'-TEMPOC-T-Phosphoramidite.

Figure 4 shows the synthesis of NINOC-T-CEP.

Figure 5 shows the synthesis of Me<sub>2</sub>NPOC-T-CEP. CEP stands for cyanoethyl N, N diisopropyl phosphoramidite.

Figure 6 shows the synthesis of Me<sub>3</sub>NPOC-T-CEP.

Figure 7 shows the synthesis of NP<sub>2</sub>NPOC-T-CEP.

Figure 8 shows the synthesis of NA<sub>1</sub>BOC-T-CEP.

### **DESCRIPTION OF THE PREFERRED EMBODIMENT**

The following definitions are set forth to illustrate and define the meaning and scope of the various terms used to describe the invention herein.

The term "alkyl" refers to a branched or straight chain acyclic, monovalent saturated hydrocarbon radical of one to twenty carbon atoms.

The term "alkenyl" refers to an unsaturated hydrocarbon radical which contains at least one carbon-carbon double bond and includes straight chain, branched chain and cyclic radicals.

The term "alkynyl" refers to an unsaturated hydrocarbon radical which contains at least one carbon-carbon triple bond and includes straight chain, branched chain and cyclic radicals.

The term "aryl" refers to an aromatic monovalent carbocyclic radical having a single ring (e.g., phenyl) or two condensed rings (e.g., naphthyl), which can optionally be

mono-, di-, or tri-substituted, independently, with alkyl, lower-alkyl, cycloalkyl, hydroxylower-alkyl, aminolower-alkyl, hydroxyl, thiol, amino, halo, nitro, lower-alkylthio, lower-alkoxy, mono-lower-alkylamino, di-lower-alkylamino, acyl, hydroxycarbonyl, lower-alkoxycarbonyl, hydroxysulfonyl, lower-alkoxysulfonyl, lower-alkylsulfonyl, lower-alkylsulfinyl, trifluoromethyl, cyano, tetrazoyl, carbamoyl, lower-alkylcarbamoyl, and di-lower-alkylcarbamoyl. Alternatively, two adjacent positions of the aromatic ring may be substituted with a methylenedioxy or ethylenedioxy group. Typically, electron-donating substituents are preferred.

The term "heteroaromatic" refers to an aromatic monovalent mono- or polycyclic radical having at least one heteroatom within the ring, *e.g.*, nitrogen, oxygen or sulfur, wherein the aromatic ring can optionally be mono-, di- or tri-substituted, independently, with alkyl, lower-alkyl, cycloalkyl, hydroxylower-alkyl, aminolower-alkyl, hydroxyl, thiol, amino, halo, nitro, lower-alkylthio, lower-alkoxy, mono-lower-alkylamino, di-lower-alkylamino, acyl, hydroxycarbonyl, lower-alkoxycarbonyl, hydroxysulfonyl, lower-alkoxysulfonyl, lower-alkylsulfonyl, lower-alkylsulfinyl, trifluoromethyl, cyano, tetrazoyl, carbamoyl, lower-alkylcarbamoyl, and di-lower-alkylcarbamoyl. For example, typical heteroaryl groups with one or more nitrogen atoms are tetrazoyl, pyridyl (*e.g.*, 4-pyridyl, 3-pyridyl, 2-pyridyl), pyrrolyl (*e.g.*, 2-pyrrolyl, 2-(N-alkyl)pyrrolyl), pyridazinyl, quinolyl (*e.g.* 2-quinolyl, 3-quinolyl etc.), imidazolyl, isoquinolyl, pyrazolyl, pyrazinyl, pyrimidinyl, pyridonyl or pyridazinonyl; typical oxygen heteroaryl radicals with an oxygen atom are 2-furyl, 3-furyl or benzofuranyl; typical sulfur heteroaryl radicals are thienyl, and benzothienyl; typical mixed heteroatom heteroaryl radicals are furazanyl and phenothiazinyl. Further the term also includes instances where a heteroatom within the ring has been oxidized, such as, for example, to form an N-oxide or sulfone.

The term "optionally substituted" refers to the presence or lack thereof of a substituent on the group being defined. When substitution is present the group may be mono-, di- or tri-substituted, independently, with alkyl, lower-alkyl, cycloalkyl, hydroxylower-alkyl, aminolower-alkyl, hydroxyl, thiol, amino, halo, nitro, lower-alkylthio, lower-alkoxy, mono-lower-alkylamino, di-lower-alkylamino, acyl, hydroxycarbonyl, lower-alkoxycarbonyl, hydroxysulfonyl, lower-alkoxysulfonyl, lower-alkylsulfonyl, lower-alkylsulfinyl, trifluoromethyl, cyano, tetrazoyl, carbamoyl, lower-alkylcarbamoyl, and di-lower-alkylcarbamoyl. Typically, electron-donating substituents such as alkyl, lower-alkyl, cycloalkyl, hydroxylower-alkyl, aminolower-alkyl, hydroxyl, thiol, amino, halo,

lower-alkylthio, lower-alkoxy, mono-lower-alkylamino and di-lower-alkylamino are preferred.

The term "electron donating group" refers to a radical group that has a lesser affinity for electrons than a hydrogen atom would if it occupied the same position in the molecule. For example, typical electron donating groups are hydroxy, alkoxy (e.g. methoxy), amino, alkylamino and dialkylamino.

The term "leaving group" means a group capable of being displaced by a nucleophile in a chemical reaction, for example halo, nitrophenoxy, pentafluorophenoxy, alkyl sulfonates (e.g., methanesulfonate), aryl sulfonates, phosphates, sulfonic acid, sulfonic acid salts, and the like.

"Activating group" refers to those groups which, when attached to a particular functional group or reactive site, render that site more reactive toward covalent bond formation with a second functional group or reactive site. The group of activating groups which are useful for a carboxylic acid include simple ester groups and anhydrides. The ester groups include alkyl, aryl and alkenyl esters and in particular such groups as 4-nitrophenyl, N-hydroxysuccinimide and pentafluorophenol. Other activating groups are known to those of skill in the art.

"Chemical library" or "array" is an intentionally created collection of differing molecules which can be prepared either synthetically or biosynthetically and screened for biological activity in a variety of different formats (e.g., libraries of soluble molecules; and libraries of compounds tethered to resin beads, silica chips, or other solid supports). The term is also intended to refer to an intentionally created collection of stereoisomers.

"Predefined region" refers to a localized area on a solid support which is, was, or is intended to be used for formation of a selected molecule and is otherwise referred to herein in the alternative as a "selected" region. The predefined region may have any convenient shape, e.g., circular, rectangular, elliptical, wedge-shaped, etc. For the sake of brevity herein, "predefined regions" are sometimes referred to simply as "regions." In some embodiments, a predefined region and, therefore, the area upon which each distinct compound is synthesized smaller than about 1 cm<sup>2</sup> or less than 1 mm<sup>2</sup>. Within these regions, the molecule synthesized therein is preferably synthesized in a substantially pure form. In additional embodiments, a predefined region can be achieved by physically separating the regions (i.e., beads, resins, gels, etc.) into wells, trays, etc.



1 "Solid support", "support", and "substrate" refer to a material or group of  
2 materials having a rigid or semi-rigid surface or surfaces. In many embodiments, at least  
3 one surface of the solid support will be substantially flat, although in some embodiments it  
4 may be desirable to physically separate synthesis regions for different compounds with, for  
5 example, wells, raised regions, pins, etched trenches, or the like. According to other  
6 embodiments, the solid support(s) will take the form of beads, resins, gels, microspheres, or  
7 other geometric configurations.

8 Isolation and purification of the compounds and intermediates described  
9 herein can be effected, if desired, by any suitable separation or purification procedure such  
10 as, for example, filtration, extraction, crystallization, column chromatography, thin-layer  
11 chromatography, thick-layer (preparative) chromatography, distillation, or a combination of  
12 these procedures. Specific illustrations of suitable separation and isolation procedures can  
13 be had by references to the examples hereinbelow. However, other equivalent separation or  
14 isolation procedures can, of course, also be used.

15 A "channel block" is a material having a plurality of grooves or recessed  
16 regions on a surface thereof. The grooves or recessed regions may take on a variety of  
17 geometric configurations, including but not limited to stripes, circles, serpentine paths, or the  
18 like. Channel blocks may be prepared in a variety of manners, including etching silicon  
19 blocks, molding or pressing polymers, etc.

20 This invention provides novel compounds which are useful for providing  
21 protecting groups in chemical synthesis, preferably in the solid phase synthesis of  
22 oligonucleotides and polypeptides and high density arrays thereof. These compounds are  
23 generally photolabile and comprise protecting groups which can be removed by photolysis to  
24 unmask a reactive group. Specifically, the preferred compounds are shown in Figures 1 and  
25 9. More specifically, the preferred compounds have R or R1 groups which can be H,  
26 optionally substituted alkyl, alkenyl, alkynyl, aryl, or heteroaromatic groups.

27 Preferably, M will be a monomeric building block that can be used to make a  
28 macromolecule. Such building blocks include amino acids, nucleic acids, nucleotides,  
29 nucleosides, monosaccharides and the like. Preferred nucleosides are deoxyadenosine,  
30 deoxycytidine, thymidine and deoxyguanosine as well as oligonucleotides incorporating  
31 such nucleosides. Preferably, the building block is linked to the photolabile protecting group  
32 via a hydroxy or amine group. When nucleotide and oligonucleotide compositions are used,  
33 with the protecting groups of this invention, the protecting groups are preferably

1 incorporated into the 3'-OH or the 5'-OH of the nucleoside. Other preferred compounds are  
2 protected peptides, proteins, oligonucleotides and oligodeoxynucleotides. Small organic  
3 molecules, proteins, hormones, antibodies and other such species having nucleophilic  
4 reactive groups can be protected using the protecting groups disclosed herein.

5 The use of nucleoside and nucleotide analogs is also contemplated by this  
6 invention to provide oligonucleotide or oligonucleoside analogs bearing the protecting  
7 groups disclosed herein. Thus the terms nucleoside, nucleotide, deoxynucleoside and  
8 deoxynucleotide generally include analogs such as those described herein. These analogs  
9 are those molecules having some structural features in common with a naturally occurring  
10 nucleoside or nucleotide such that when incorporated into an oligonucleotide or  
11 oligonucleoside sequence, they allow hybridization with a naturally occurring  
12 oligonucleotide sequence in solution. Typically, these analogs are derived from naturally  
13 occurring nucleosides and nucleotides by replacing and/or modifying the base, the ribose or  
14 the phosphodiester moiety. The changes can be tailor made to stabilize or destabilize hybrid  
15 formation or enhance the specificity of hybridization with a complementary nucleic acid  
16 sequence as desired.

17 Analogs also include protected and/or modified monomers as are  
18 conventionally used in oligonucleotide synthesis. As one of skill in the art is well aware  
19 oligonucleotide synthesis uses a variety of base-protected deoxynucleoside derivatives in  
20 which one or more of the nitrogens of the purine and pyrimidine moiety are protected by  
21 groups such as dimethoxytrityl, benzyl, tert-butyl, isobutyl and the like. Specific monomeric  
22 building blocks which are encompassed by this invention include base protected  
23 deoxynucleoside H-phosphonates and deoxynucleoside phosphoramidites.

24 For instance, structural groups are optionally added to the ribose or base of a  
25 nucleoside for incorporation into an oligonucleotide, such as a methyl, propyl or allyl group  
26 at the 2'-O position on the ribose, or a fluoro group which substitutes for the 2'-O group, or a  
27 bromo group on the ribonucleoside base. 2'-O-methyloligoribonucleotides (2'-O-MeORNs)  
28 have a higher affinity for complementary nucleic acids (especially RNA) than their  
29 unmodified counterparts. 2'-O-MeORNA phosphoramidite monomers are available  
30 commercially, *e.g.*, from Chem Genes Corp. or Glen Research, Inc. Alternatively,  
31 deazapurines and deazapyrimidines in which one or more N atoms of the purine or  
32 pyrimidine heterocyclic ring are replaced by C atoms can also be used.

1           The phosphodiester linkage, or "sugar-phosphate backbone" of the  
2 oligonucleotide analogue can also be substituted or modified, for instance with methyl  
3 phosphonates or O-methyl phosphates. Another example of an oligonucleotide analogue for  
4 purposes of this disclosure includes "peptide nucleic acids" in which a polyamide backbone  
5 is attached to oligonucleotide bases, or modified oligonucleotide bases. Peptide nucleic  
6 acids which comprise a polyamide backbone and the bases found in naturally occurring  
7 nucleosides are commercially available.

8           Nucleotides with modified bases can also be used in this invention. Some  
9 examples of base modifications include 2-aminoadenine, 5-methylcytosine, 5-(propyn-1-  
10 yl)cytosine, 5-(propyn-1-yl)uracil, 5-bromouracil, and 5-bromocytosine which can be  
11 incorporated into oligonucleotides in order to increase binding affinity for complementary  
12 nucleic acids. Groups can also be linked to various positions on the nucleoside sugar ring or  
13 on the purine or pyrimidine rings which may stabilize the duplex by electrostatic interactions  
14 with the negatively charged phosphate backbone, or through hydrogen bonding interactions  
15 in the major and minor grooves. For example, adenosine and guanosine nucleotides can be  
16 substituted at the N<sup>2</sup> position with an imidazolyl propyl group, increasing duplex stability.  
17 Universal base analogues such as 3-nitropyrrole and 5-nitroindole can also be included. A  
18 variety of modified oligonucleotides and oligonucleotide analogs suitable for use in this  
19 invention are described "Antisense Research and Applications", S.T. Crooke and B. LeBleu  
20 (eds.) (CRC Press, 1993) and "Carbohydrate Modifications in Antisense Research" in ACS  
21 Symp. Ser. #580, Y.S. Sanghvi and P.D. Cook (eds.) ACS, Washington, D.C. 1994).

22           Compounds of this invention can be prepared by carbonylating an alcohol or  
23 amine precursor "Y" with a carbonylation reagent such as for example, phosgene (COCl<sub>2</sub>),  
24 carbonyldiimidazole or pentafluorophenoxy chloroformate and the like to provide Y-C(O)-X  
25 where X is a leaving group derived from the carbonylating reagent (Cl, if phosgene was  
26 used, pentafluorophenoxy, if pentafluorophenoxy chloroformate was used, etc.). This  
27 intermediate, Y-C(O)-X is then reacted with a molecule M carrying a nucleophilic group  
28 whose protection is desired to yield a protected building block Y-C(O)-M.

29           Alternatively, one may first carbonylate the group on the molecule being  
30 protected with a carbonylation reagent, such as one described above, and subsequently  
31 displace the leaving group X thus inserted with the hydroxyl group of the aromatic carbinol.  
32 In either procedure, one frequently uses a base such as triethylamine or  
33 diisopropylethylamine and the like to facilitate the displacement of the leaving group.

One of skill in the art will recognize that the protecting groups disclosed herein can also be attached to species not traditionally considered as "molecules". Therefore, compositions such as solid surfaces (*e.g.*, paper, nitrocellulose, glass, polystyrene, silicon, modified silicon, GaAs, silica and the like), gels (*e.g.*, agarose, sepharose, polyacrylamide and the like) to which the protecting groups disclosed herein are attached are also contemplated by this invention.

The protecting groups of this invention are typically removed by photolysis, *i.e.* by irradiation, though in selected cases it may be advantageous to use acid or base catalyzed cleavage conditions. The synthesis can occur in either the 3'→5' or 5'→3' directions. Generally irradiation is at wavelengths greater than about 350 nm, preferably at about 365 nm. The photolysis is usually conducted in the presence of hydroxylic solvents, such as aqueous, alcoholic or mixed aqueous-alcoholic or mixed aqueous-organic solvent mixtures. Alcoholic solvents frequently used include methanol and ethanol. The photolysis medium may also include nucleophilic scavengers such as hydrogen peroxide. Photolysis is frequently conducted at neutral or basic pH.

This invention also provides a method of attaching a molecule with a reactive site to a support, comprising the steps of:

- (a) providing a support with a reactive site;
- (b) binding a molecule to the reactive site, said first molecule comprising a masked reactive site attached to a photolabile protecting group of the formula Y-C(O)-, and
- (c) removing the photolabile protecting group to provide a derivatized support comprising the molecule with an unmasked reactive site immobilized thereon.

As one of skill will recognize, the process can be repeated to generate a compound comprising a chain of component molecules attached to the solid support. In a "mix and match" approach, the photolabile protecting groups may be varied at different steps in the process depending on the ease of synthesis of the protected precursor molecule. Alternatively, photolabile protecting groups can be used in some steps of the synthesis and chemically labile (*e.g.* acid or base sensitive groups) can be used in other steps, depending for example on the availability of the component monomers, the sensitivity of the substrate and the like. This method can also be generalized to be used in preparing arrays of compounds, each compound being attached to a different and identifiable site on the support as is disclosed in U.S. Patent Nos. 5,143,854, 5,384,261, 5,424,186, 5,445,934, 6,022,963 and

1 copending U.S. Patent Application, Serial No. 08/376,963, filed January 23, 1995,  
2 incorporated for reference for all purposes in their entirety.

3 Thus, a related aspect of this invention provides a method of forming, from  
4 component molecules, a plurality of compounds on a support, each compound occupying a  
5 separate region of the support, said method comprising the steps of:

- 6 (a) activating a region of the support;  
7 (b) binding a molecule to the region, said molecule comprising a masked  
8 reactive site linked to a photolabile protecting group of the formula Y-C(O)-, and  
9 (c) repeating steps (a) and (b) on other regions of the support whereby each  
10 of said other regions has bound thereto another molecule comprising a masked reactive site  
11 linked to the photolabile protecting group, wherein said another molecule may be the same  
12 or different from that used in step (b);  
13 (d) removing the photolabile protecting group from one of the molecules  
14 bound to one of the regions of the support to provide a region bearing a molecule with an  
15 unmasked reactive site;  
16 (e) binding an additional molecule to the molecule with an unmasked  
17 reactive site;  
18 (f) repeating steps (d) and (e) on regions of the support until a desired  
19 plurality of compounds is formed from the component molecules, each compound occupying  
20 separate regions of the support.

21 A related method of forming a plurality of compounds on predefined regions  
22 of a support involves binding a molecule with a reactive site protected with a chemically  
23 labile protecting group to an activated region of the support and chemically removing the  
24 chemically labile protecting group to reveal the reactive site. The reactive site is then  
25 protected with a photolabile protecting group of this invention. This process is repeated for  
26 other regions of the support with other molecules as desired to provide a support having  
27 molecules with reactive sites protected by photolabile protecting groups on separate regions  
28 of the support. Reactive sites can be unmasked by removing the photolabile group from  
29 selected regions and coupled to additional molecules with photolabile protecting groups as  
30 described earlier to build up arrays of compounds on the support. Again, in a "mix and  
31 match" approach, monomers with chemically labile protecting groups can be attached to a  
32 reactive site on the substrate (i.e., on the support itself when the first layer of monomers is  
33 being assembled or subsequently onto an already attached monomer whose reactive site has

1 been unmasked) and these chemically labile protecting groups can be replaced by a  
2 photolabile protecting groups of this invention. The replacement is accomplished by  
3 removing the chemically labile protecting group under conditions that do not affect any  
4 photolabile groups which may be on the support. This then reveals an unmasked reactive  
5 site on the monomer which had carried the chemically labile protecting group and this  
6 unmasked reactive site is reacted with a reagent of the formula  $Y-C(O)-X$ , where X is a  
7 leaving group. Thereby, this region of the support is protected by a photolabile protecting  
8 group which can be selectively removed by light directed systems described in U.S. Patent  
9 Nos. 5,143,854, 5,384,261, 5,424,186 and 5,445,934 and further described below  
10 (incorporated by reference in their entireties for all purposes). This method is particularly  
11 useful when the monomers are more readily available carrying chemically labile protecting  
12 groups than the photolabile protecting groups described herein. It will be recognized that  
13 any method of forming a chain of compounds or an array of compounds on a support using  
14 in at least one step a protecting group/reagent or compound of this invention is within the  
15 scope of the methods this invention.

16 Generally, these methods involve sequential addition of monomers to build  
17 up an array of polymeric species on a support by activating predefined regions of a substrate  
18 or solid support and then contacting the substrate with a protected monomer of this invention  
19 (e.g., a protected nucleoside or amino acid). It will be recognized that the individual  
20 monomers can be varied from step to step. A common support is a glass or silica substrate  
21 as is used in semiconductor devices.

22 The predefined regions can be activated with a light source, typically shown through  
23 a screen such as a photolithographic mask similar to the techniques used in integrated circuit  
24 fabrication. Other regions of the support remain inactive because they are blocked by the  
25 mask from illumination and remain chemically protected. Thus, a light pattern defines  
26 which regions of the support react with a given monomer. The protected monomer reacts  
27 with the activated regions and is immobilized therein. The protecting group is removed by  
28 photolysis and washed off with unreacted monomer. By repeatedly activating different sets  
29 of predefined regions and contacting different monomer solutions with the substrate, a  
30 diverse array of polymers of known composition at defined regions of the substrate can be  
31 prepared. Arrays of  $10^6$ ,  $10^7$ ,  $10^8$ ,  $10^9$ ,  $10^{10}$ ,  $10^{11}$ ,  $10^{12}$  or more different polymers can be  
32 assembled on the substrate. The regions may be  $1\text{ mm}^2$  or larger, typically  $10\text{ }\mu\text{m}^2$  and may  
33 be as small as  $1\text{ }\mu\text{m}^2$ .

1           The methods described herein may also employ component molecules  
2 comprising a masked reactive site attached to a photolabile protecting group having the  
3 structure Y. In such cases, the protecting group is attached to an acidic reactive site, such as  
4 a carboxylate or phosphate and is removed by photolysis.

5           The solid substrate or solid support may be of any form, although they  
6 preferably will be planar and transparent (and potentially some three dimensional structure).  
7 The supports need not necessarily be homogenous in size, shape or composition, although  
8 the supports usually and preferably will be uniform. In some embodiments, supports that are  
9 very uniform in size may be particularly preferred. In another embodiment, two or more  
10 distinctly different populations of solid supports may be used for certain purposes.

11           Solid supports may consist of many materials, limited primarily by capacity  
12 for derivatization to attach any of a number of chemically reactive groups and compatibility  
13 with the synthetic chemistry used to produce the array and, in some embodiments, the  
14 methods used for tag attachment and/or synthesis. Suitable support materials typically will  
15 be the type of material commonly used in peptide and polymer synthesis and include glass,  
16 latex, heavily cross-linked polystyrene or similar polymers, gold or other colloidal metal  
17 particles, and other materials known to those skilled in the art. The chemically reactive  
18 groups with which such solid supports may be derivatized are those commonly used for solid  
19 phase synthesis of the polymer and thus will be well known to those skilled in the art, *i.e.*,  
20 carboxyls, amines, and hydroxyls.

21           To improve washing efficiencies, one can employ nonporous supports or  
22 other solid supports less porous than typical peptide synthesis supports; however, for certain  
23 applications of the invention, quite porous beads, resins, or other supports work well and are  
24 often preferable. One such support is a resin in the form of beads. In general, the bead size  
25 is in the range of 1 nm to 100  $\mu$ m, but a more massive solid support of up to 1 mm in size  
26 may sometimes be used. Particularly preferred resins include Sasrin resin (a polystyrene  
27 resin available from Bachem Bioscience, Switzerland); and TentaGel S AC, TentaGel PHB,  
28 or TentaGel S NH<sub>2</sub> resin (polystyrene-polyethylene glycol copolymer resins available from  
29 Rappe Polymere, Tubingen, Germany). Other preferred supports are commercially available  
30 and described by Novabiochem, La Jolla, California.

31           In other embodiments, the solid substrate is flat, or alternatively, may take on  
32 alternative surface configurations. For example, the solid substrate may contain raised or  
33 depressed regions on which synthesis takes place. In some embodiments, the solid substrate

will be chosen to provide appropriate light-absorbing characteristics. For example, the substrate may be a polymerized Langmuir Blodgett film, functionalized glass, Si, Ge, GaAs, GaP, SiO<sub>2</sub>, SiN<sub>4</sub>, modified silicon, or any one of a variety of gels or polymers such as (poly)tetrafluorethylene, (poly)vinylidenedifluoride, polystyrene, polycarbonate, or combinations thereof. Other suitable solid substrate material will be readily apparent to those of skill in the art. Preferably, the surface of the solid substrate will contain reactive groups, which could be carboxyl, amino, hydroxyl, thiol, or the like. More preferably, the surface will be optically transparent and will have surface Si-OH functionalities, such as are found on silica surfaces.

The photolabile protecting groups and protected monomers disclosed herein can also be used in bead based methods of immobilization of arrays of molecules on solid supports.

A general approach for bead based synthesis is described in copending application Serial Nos. 07/762,522 (filed September 18, 1991); 07/946,239 (filed September 16, 1992); 08/146,886 (filed November 2, 1993); 07/876,792 (filed April 29, 1992) and PCT/US93/04145 (filed April 28, 1993), Lam et al. (1991) *Nature* **354**:82-84; PCT application no. 92/00091 and Houghten et al, (1991) *Nature* **354**:84-86, each of which is incorporated herein by reference for all purposes.

Other methods of immobilization of arrays of molecules in which the photocleavable protecting groups of this invention can be used include pin based arrays and flow channel and spotting methods.

Photocleavable arrays also can be prepared using the pin approach developed by Geysen et al. for combinatorial solid-phase peptide synthesis. A description of this method is offered by Geysen et al., *J. Immunol. Meth.* (1987) **102**:259-274, incorporated herein by reference.

Additional methods applicable to library synthesis on a single substrate are described in U.S. Patent Nos. 5,384,261, 5,677,195, 6,040,193 that are hereby incorporated by reference in their entireties for all purposes. In the methods disclosed in these applications, reagents are delivered to the substrate by either (1) flowing within a channel defined on predefined regions or (2) "spotting" on predefined regions. However, other approaches, as well as combinations of spotting and flowing, may be employed. In each instance, certain activated regions of the substrate are mechanically separated from other regions when the monomer solutions are delivered to the various reaction sites.



Photocleavable linkers are particularly suitable for this technology as this delivery method may otherwise result in poor synthesis fidelity due to spreading, reagent dilution, inaccurate delivery, and the like. By using a photocleavable linker, rather than a conventional acid-cleavable linker, the purest material can be selectively cleaved from the surface for subsequent assaying or other procedures. More specifically, masks can be used when cleaving the linker to ensure that only linker in the center of the delivery area (*i.e.*, the area where reagent delivery is most consistent and reproducible) is cleaved. Accordingly, the material thus selectively cleaved will be of higher purity than if the material were taken from the entire surface.

Typically, the molecules used in this method will be the monomeric components of complex macromolecules. These monomeric components can be small ligand molecules, amino acids, nucleic acids, nucleotides, nucleosides, monosaccharides and the like, thereby allowing one to synthesize arrays of complex macromolecules or polymeric sequences, such as polypeptides, nucleic acids and synthetic receptors, on the solid support.

#### EXAMPLE

Examples of the preferred groups shown in Figure 2 were synthesized and tested as 5'-photolabile protecting groups on thymidine phosphoramidite monomers. Surface photolysis rates in different solvents (std. 365nm lightsource) were determined as described elsewhere (McGall et al., JACS 1997, 119: 5081, hereby incorporated by reference in its entirety for all purposes). Standard coupling efficiency measurements were made using the cleavable linker HPLC analysis technique (see U.S.S.No. 09/545,207, and attorney docket no. 3233.1, which are both hereby incorporated by reference in their entireties).

Figure 1 shows the preferred compounds and their synthesis. It shows the general structures of the preferred structures, the preferred structures, their synthesis, the yields of the nucleic acid sequences formed using the preferred protecting groups, and the photolysis conditions. Also, the synthesis steps are annotated with references that relate to the specific synthesis. All of these references are hereby incorporated by reference in their entireties for all purposes.

5'-TEMPOC-T-Phosphoramidite was synthesized using the steps outlined in Fig. 3 and the details shown in the references in that Figure. Specifically, the following references are hereby incorporated by reference in their entireties for all purposes as well as the steps that are cited: Dyer, et al. JOC 64:7988 (1999); Tetrahedron Lett., 38(52), 8933-4 (1997);

1 Mcgall, et al., JACS 119:5081 (1997). The Fig. indicates that triphosgene may work equally  
2 well for step #1 and that chloroformate could probably be used without purification in step  
3 #2. NINOC-T-CEP was synthesized according to the steps shown in Fig. 4 and the  
4 following references are incorporated by reference in their entireties for all purposes as well  
5 as the steps that are cited; Bromidge, et al. (1998) J. Med. Chem. 41: 1598; Brooker, LS, et  
6 al. (1953) U.S. Patent No. 2,646,430; Boekelheide, et al. (1954) J. Org. Chem. 19: 504;  
7 Bennet, et al. (1941) J. Chem. Soc. 74:244; and Mortensen, et al. (1996) Org. Prep. Proc. Int.  
8 28: 123. Figs. 5-8 show the synthesis of the following compounds; Me2NPOC-T-CEP;  
9 Me3NPOC-T-CEP; and NA1BOC-T-CEP. Fig. 8 refers to Aust. J. Chem 48:1969-70 which  
10 is also incorporated by reference in its entirety. Abbreviations used in the first step of the  
11 processes indicate the source of the material. For example, DAV is Davos, LAN is  
12 Lancaster, ALH is Adrich. CEP stands for cyanoethyl N, N diisopropyl phosphoramidite.

13 The foregoing invention has been described in some detail by way of illustration and  
14 examples, for purposes of clarity and understanding. It will be obvious to one of skill in the  
15 art that changes and modifications may be practiced within the scope of the appended  
16 claims. Therefore, it is to be understood that the above description is intended to be  
17 illustrative and not restrictive. The scope of the invention should, therefore, be determined  
18 not with reference to the above description, but should instead be determined with reference  
19 to the following appended claims, along with the full scope of equivalents to which such  
20 claims are entitled.

21 All patents, patent applications and publications cited in this application are  
22 hereby incorporated by reference in their entirety for all purposes to the same extent as if  
23 each individual patent, patent application or publication were so individually denoted.

Seven chemical structures are shown, representing various nitro-substituted compounds. The structures are as follows:

- Top left: A benzene ring with a nitro group ( $\text{NO}_2$ ) at the 1-position and a thioether group ( $-\text{S}-\text{CH}_2-\text{CH}_2-\text{O}-\text{C}(=\text{O})-\text{CH}_3$ ) at the 2-position. A handwritten label  $\text{R1}$  is above the thioether group.
- Top middle: A benzene ring with a nitro group ( $\text{NO}_2$ ) at the 1-position and an amide group ( $-\text{N}(\text{R1})-\text{C}(=\text{O})-\text{CH}_3$ ) at the 2-position. A handwritten label  $\text{R1}$  is above the amide group.
- Top right: A benzene ring with a nitro group ( $\text{NO}_2$ ) at the 1-position and an ester group ( $-\text{O}-\text{C}(=\text{O})-\text{CH}_3$ ) at the 2-position.
- Middle left: A benzene ring with a nitro group ( $\text{NO}_2$ ) at the 1-position and an ester group ( $-\text{O}-\text{C}(=\text{O})-\text{CH}_3$ ) at the 3-position.
- Middle: A benzene ring with a nitro group ( $\text{NO}_2$ ) at the 1-position and a thioether group ( $-\text{S}-\text{CH}_2-\text{CH}_2-\text{O}-\text{C}(=\text{O})-\text{CH}_3$ ) at the 2-position.
- Middle right: A naphthalene ring system with a nitro group ( $\text{NO}_2$ ) at the 1-position and a methyl ester group ( $-\text{CH}(\text{Me})-\text{O}-\text{C}(=\text{O})-\text{O}-\text{CH}_3$ ) at the 2-position.
- Bottom left: A naphthalene ring system with a nitro group ( $\text{NO}_2$ ) at the 1-position and an ester group ( $-\text{O}-\text{C}(=\text{O})-\text{CH}_3$ ) at the 2-position. Handwritten labels  $\text{A}$  and  $\text{B}$  are present near the ester group.
- Bottom right: A benzene ring with a nitro group ( $\text{NO}_2$ ) at the 1-position and a thioether group ( $-\text{S}-\text{CH}_2-\text{CH}_2-\text{O}-\text{C}(=\text{O})-\text{CH}_3$ ) at the 2-position. Handwritten labels  $\text{A}$  and  $\text{B}$  are present near the thioether group.

(a) providing a support with a reactive site;

(b) binding a molecule to the reactive site, said first molecule comprising a masked reactive site attached to a photolabile protecting group of the formula  $Y-C(O)-$ , wherein Y is a chemical group as claimed in claim 1;

(c) removing the photolabile protecting group to provide a derivatized support comprising the molecule with an unmasked reactive site immobilized thereon.

6. The method of Claim 5, wherein the binding step in (b) is covalent.

7. The method of Claim 5, further comprising:

(a) coupling a second molecule to the unmasked reactive site, which second molecule comprises a second masked reactive site attached to the photolabile protecting group to produce a derivatized support having immobilized thereon a chain of the first and second molecules; and

(b) removing the photolabile protecting group to provide a derivatized support with a chain of the first and second molecules with a second unmasked reactive site immobilized thereon.

8. The method of Claim 5, further comprising repeating steps (a) and (b) of Claim 10 with a succession of molecules to provide a chain of molecules immobilized on the support.

9. The method of Claim 5, wherein the molecules are deoxynucleosides.

10. The method of Claim 5, wherein the support is a glass or silica substrate.

11. The method of Claim 9, wherein the deoxynucleosides are linked to the photolabile group via a 5'-OH.

12. The method of Claim 7, wherein the photolabile group is removed by irradiation at a wavelength of greater than 350 nm.

13. The method of Claim 12, wherein the wavelength is about 365 nm.

14. A method of forming, from component molecules, a plurality of compounds on a support, each compound occupying a separate predefined region of the support, said method comprising the steps of:

(a) activating a region of the support;

(b) binding a molecule to the first region, said molecule comprising a masked reactive site linked to a photolabile protecting group of the formula Y-C(O)-, wherein Y is a chemical compound of the structure shown in claim 1;

(c) repeating steps (a) and (b) on other regions of the support whereby each of said other regions has bound thereto another molecule comprising a masked reactive site linked to the photolabile protecting group, wherein said another molecules may be the same or different from that used in step (b);

- (d) removing the photolabile protecting group from one of the molecules bound to one of the regions of the support to provide a region bearing a molecule with an unmasked reactive site;
  - (e) binding an additional molecule to the molecule with an unmasked reactive site;
  - (f) repeating steps (d) and (e) on regions of the support until a desired plurality of compounds is formed from the component molecules, each compound occupying separate regions of the support.
15. The method of Claim 14, wherein the binding steps are covalent.
  16. The method of Claim 14, wherein the molecules are deoxynucleosides.
  17. The method of Claim 14, wherein the support is a glass or silica substrate.
  18. The method of Claim 16, wherein the deoxynucleosides are linked to the photolabile group via a 5'-OH or the 3'-OH.
  19. The method of Claim 14, wherein the photolabile group is removed by irradiation at a wavelength of greater than 350 nm.
  20. The method of Claim 19, wherein the wavelength is about 365 nm.
  21. The method of Claim 14, wherein at least  $10^6$  chains are immobilized on the support.
  22. The method of Claim 14, wherein each of the regions has an area of between about  $1 \mu\text{m}^2$  and  $10,000 \mu\text{m}^2$ .
  23. The method of Claim 14, further comprising:
    - (a) covalently binding a second molecule comprising a masked reactive site linked to a chemically labile protecting group to a reactive site, wherein the reactive site is either on an activated region of the support as formed in step (a) of Claim 19 or is an unmasked reactive site on a molecule on the support as formed in step (d) of Claim 19;
    - (b) replacing the chemically labile protecting group with the photolabile protecting group to provide a region of the support having a molecule with the photolabile protecting group; and
    - (c) repeating steps (d) - (f) of Claim 19 as desired.
  24. A compound as recited in claim 1 wherein the compound Y is Me2NPOC; Me3NPOC; NP2NPOC; NA1BOC; 5'-TEMPOC and NINOC.

25. A compound as recited in claim 4 wherein the compound Y is Me<sub>2</sub>NPOC-T-

26. A method in accordance with claim 9 wherein the compound Y is Me<sub>2</sub>NPOC;

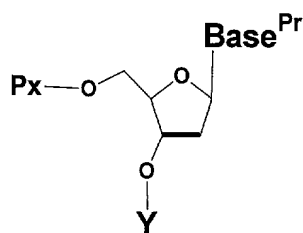
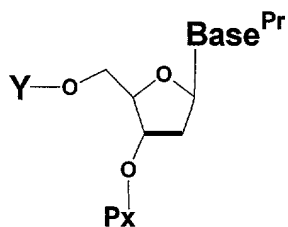
27. A method in accordance with claim 9 wherein the compound Y is Me<sub>2</sub>NPOC-T-

28. A method in accordance with claim 14 wherein the compound Y is

29. .A method in accordance with claim 16 wherein the compound Y is

Novel compounds are provided which are useful as linking groups in chemical synthesis, preferably in the solid phase synthesis of oligonucleotides and polypeptides. These compounds are generally photolabile and comprise protecting groups which can be removed by photolysis to unmask a reactive group. The protecting group has the general formula Y-C(O)-wherein: Y is a chemical structure as shown in Figure 1. Also provided is a method of forming, from component molecules, a plurality of compounds on a support, each compound occupying a separate predefined region of the support, using the protected compounds described above.

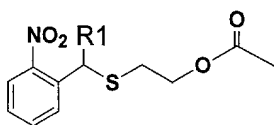
Figure 1



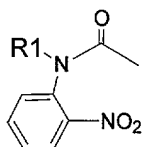
**Px = phosphoramidite, H-phosphonate, phosphate triester**

**Y = general structures (R1 = H, alkyl, aryl)**

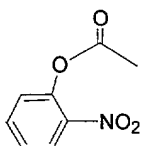
o-nitrobenzylthioethoxycarbonyl NBTEOC:



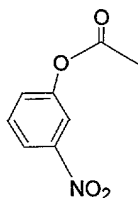
o-nitrophenylaminocarbonyl NPAC



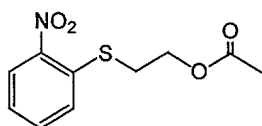
o-nitrophenoxycarbonyl N2POC



m-nitrophenoxycarbonyl N3POC



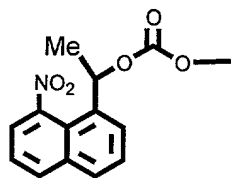
o-nitrophenylthioethoxycarbonyl



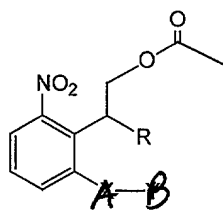


## Figure 1 (cont'd)

$\alpha$ -methyl-8-nitronaphthylmethoxycarbonyl MeNMOC



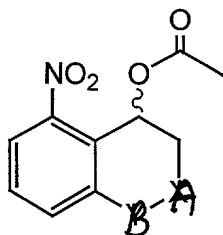
6-substituted 2-(o-nitrophenyl)-2-propyloxycarbonyl 6NPPOC



(A) = O, S, N-alkyl, N-aryl, (CH<sub>2</sub>)<sub>n</sub>, where n = 0 - ~3)

(B) = aprotic weakly basic group - eg.: N-alkylimidazole)

cyclic o-nitrobenzyloxycarbonyl



Specific examples tested...

Figure 2

TEMPOC

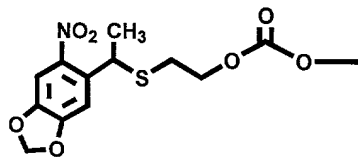
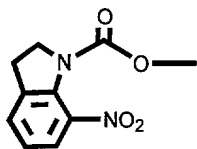
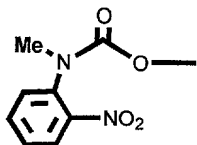


Figure 2 (cont'd)

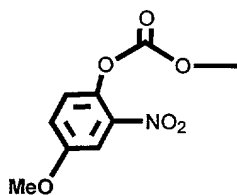
NIOC



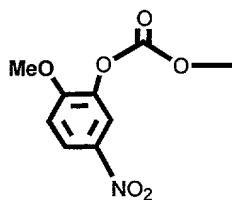
NAMOC



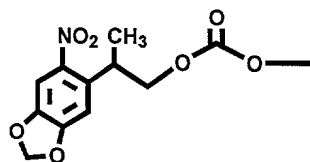
MeN2POC



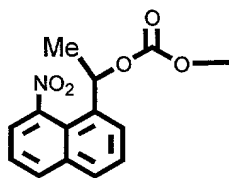
MeN3POC



NP2POC



NNEOC



# Figure 2 (cont'd)

## Coupling Efficiency Data

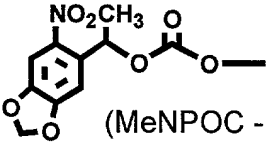
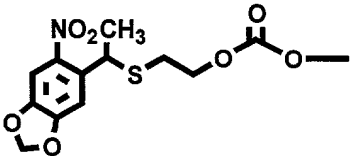
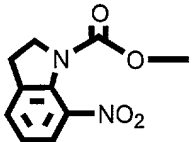
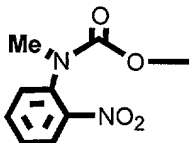
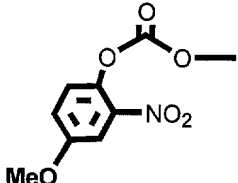
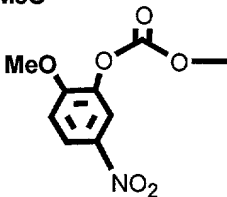
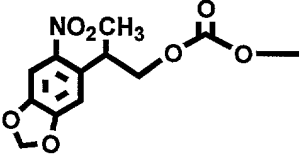
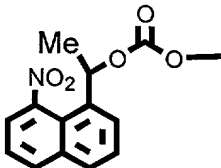
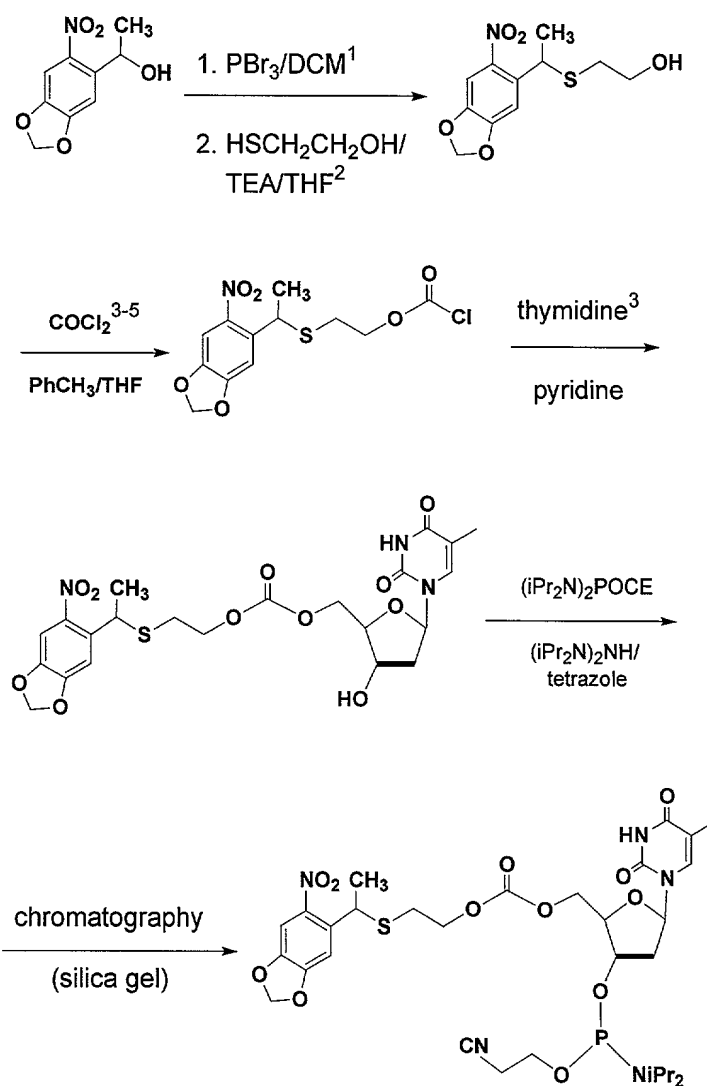
Y =	stepwise yield	photolysis conditions
 (MeNPOC - control)	~88 %	nonpolar solvent
	~85 %	MeOH
	95%	DMSO
	94%	nucleophilic solvent (MeOH)
	~80%	nucleophilic solvent (MeOH)
	~75 %	nucleophilic solvent (MeOH)
	90 %	basic solvent (1%NMI/DMSO)
	96 %	DMSO

Figure 3

### 5'-TEMPOC-T-Phosphoramidite



1 Dyer, et al. JOC 64: 7988 (1999)

2 Tetrahedron Lett., 38(52), 8933-4 (1997)

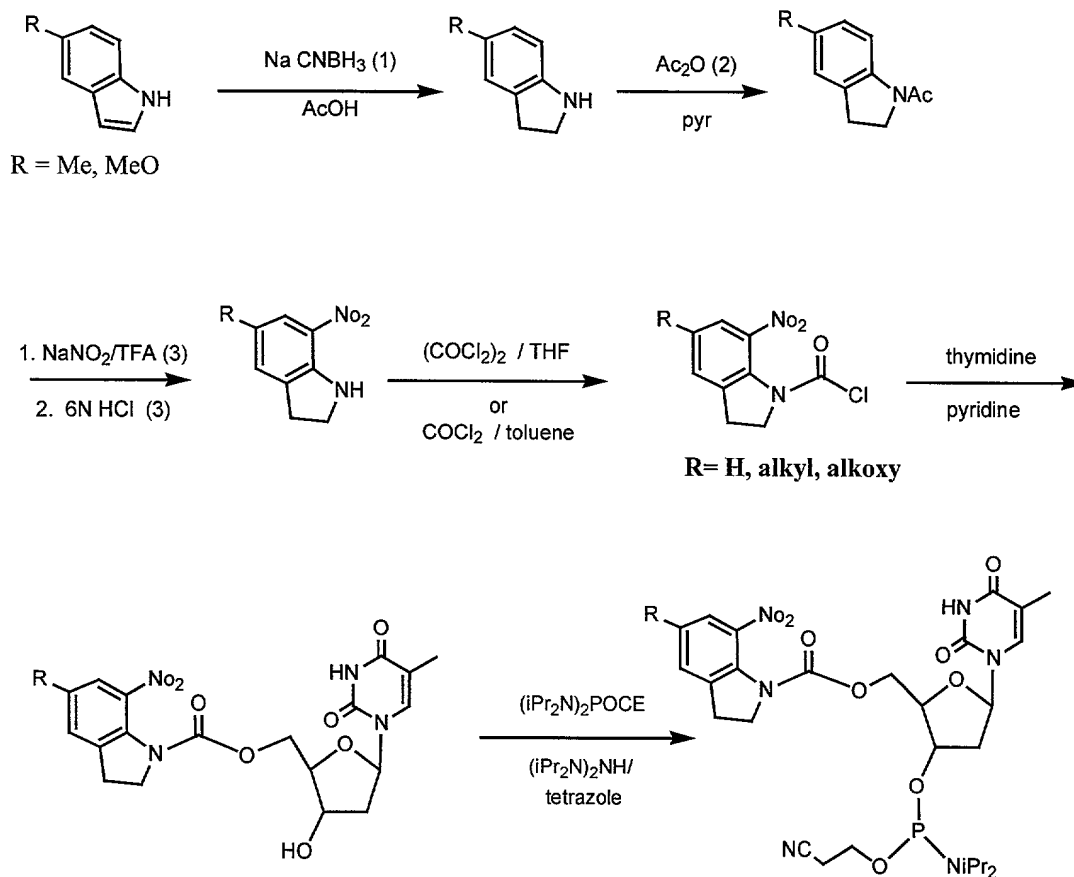
3 McGall, et al. JACS 119: 5081 (1997)

4 triphosgene may work equally well for this step.

5 chloroformate can probably be used without purification.

Figure 4

## Synthesis of NINOC-T-CEP



(1) Bromidge, et al. (1998) *J. Med. Chem.* **41**: 1598.

(2) (i) Brooker, LS, et al. (1953) *US Pat.* 2,646,430; (ii) Boekelheide, et al. (1954) *J. Org. Chem.* **19**: 504.; (iii) Bennet, et al (1941) *J. Chem. Soc.* **74**: 244.

(3) Mortensen, et al. (1996) *Org. Prep. Proc. Int.* **28**: 123.

Figure 5

Me2NPOC-T-CEP

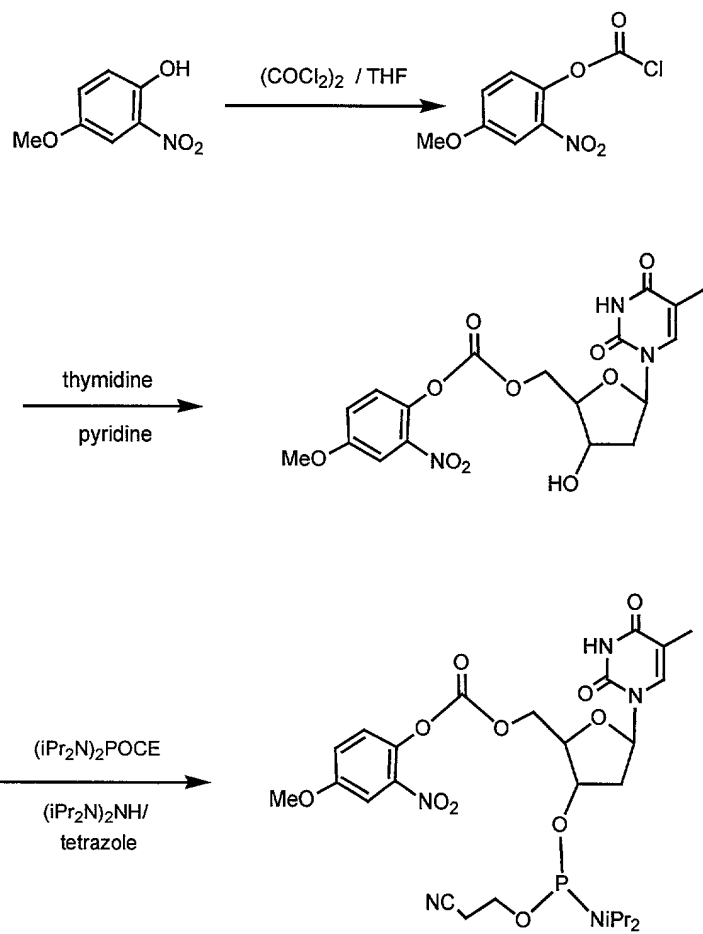


Figure 6

# Me3NPOC-T-CEP

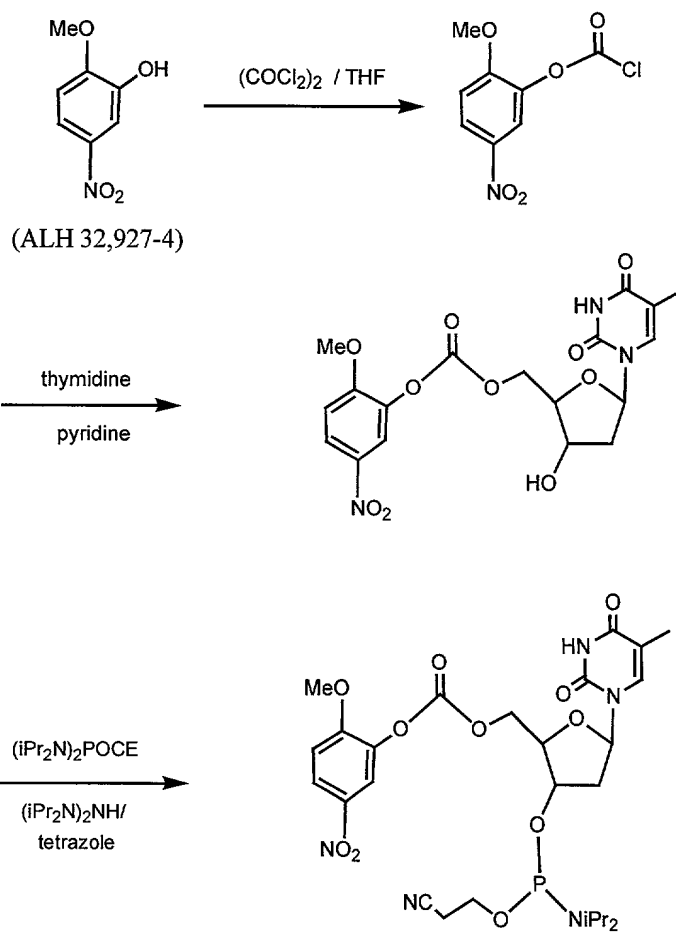


Figure 7

NP2POC-T-CEP

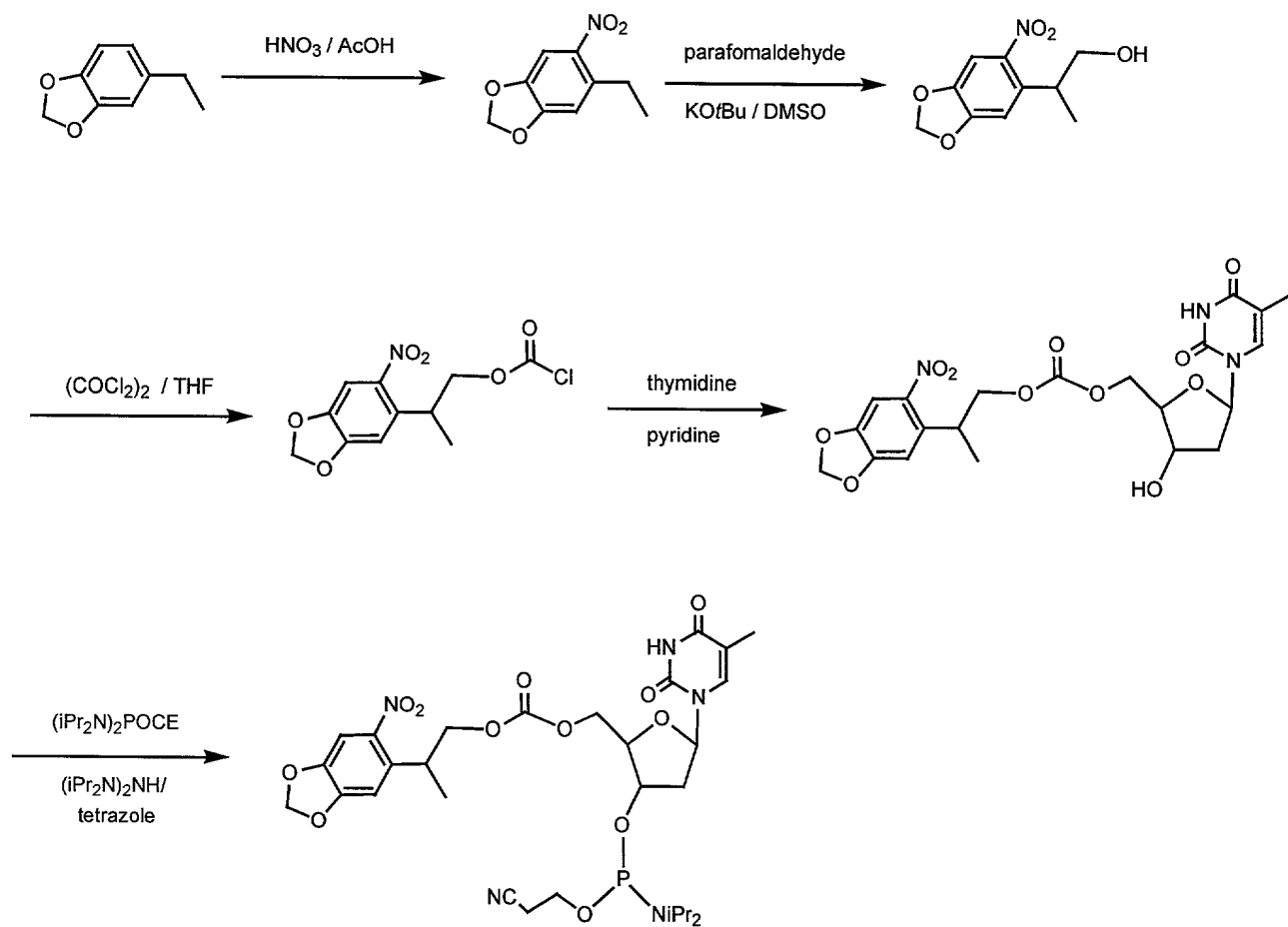




Figure 8

# NNEOC-T-CEP

CC(C)C(=O)c1ccc2ccccc2c1

